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Key Concepts 15.1 The high surface tension of water and low vapour pressure are due to the hydrogen bonding between the molecules The structure of ice is a regular open frame-work of water molecules held together by hydrogen bonds and arranged like a honeycomb Molecules in liquid water are packed closer together than they are in ice

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Kövinban, Prentice Hall Chemistry: Chapter 15. Surface Tension. Surfactant. Key Concepts 15.1. Aqueous Solution. The inward force that tends to minimize the surface area of al. Any substance that interferes with the hydrogen bonding betwee. The high surface tension of water and low vapour pressure are.

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Chapter 15 Chemical Equilibrium

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