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Octave Levenspiel was a professor of the field Chemical engineering at Oregon State University. In this vast and evergreen field his major interests lied in Chemical Reaction Engineering which is one of

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the core subjects in
Chemical Engineering. He
published this book which is
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1 Chemical reactions 1.1

Rate of reaction and

dependence on temperature We

will once again look at the

formation of ammonia (NH_3)

from nitrogen and hydrogen

(see section Chemical

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equilibrium of the
thermodynamics chapter).

This reaction follows the

equation: $N_2 + 3H_2 \rightleftharpoons 2NH_3$

(1) $H^0 = 92 \text{ kJ mol}^{-1}$ $S^0 = 192$

$\text{J mol}^{-1} \text{ K}^{-1}$ To find the Gibbs free
energy of ...

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By sizing a chemical reactor

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we mean we're either
detering the reactor volume
to achieve a given
conversion or determine the
conversion that can be
achieved in a given reactor
type and size. Here we will
assume that we will be given

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$-r_A = f(X)$ and F_{A0} . In chapter 3 we show how to find $-r_A = f(X)$. Given $-r_A$ as a function of conversion, $-r_A = f(X)$, one can size any type of reactor.

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Elements of Chemical

Reaction Engineering

Octave Levenspiel was a professor of chemical engineering at Oregon State University. His principal interest was chemical reaction engineering, and he

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was the author of a major textbook Chemical Reaction Engineering as well as numerous research publications.

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